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Highly efficient synthesis of buflavine: a unique *Amaryllidaceae* alkaloid

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Abstract—The *Amaryllidaceae* alkaloid buflavine 1 has been synthesized in three steps by Suzuki–Miyaura cross coupling, reduction and the cascade reactions of Pictet–Spengler type and Eschweiler–Clarke *N*-methylation. © 2003 Elsevier Science Ltd. All rights reserved.

Buflavine 1 belongs to a group of natural Amaryllidaceae alkaloids isolated from Boophane flava bulbs. It possesses a very rare 5,6,7,8-tetrahydrodibenz[c,e]azocine skeleton composed of a biaryl ring system linked via an eight-membered N-heterocyclic ring. Such compounds have been shown to exhibit potential α -adrenolytic and anti-serotonin activities. The unique structure of buflavine and its interesting biological activities have prompted our efforts to synthesize this alkaloid. All

previous synthetic routes^{3,4} including the most recent publication⁵ firstly involve formation of the biaryl ring by three different standard procedures (Ullmann, Suzuki–Miyaura and Meyer's biaryl coupling) followed by construction of the eight-membered *N*-heterocyclic ring via condensation of a C₁ side-chain with the remaining side-chain. Herein, we are pleased to report an efficient three-step total synthesis of buflavine 1 based on the retrosynthetic analysis outlined in Scheme 1.

$$\begin{array}{c} \text{MeO} \\ \text{MeO$$

Scheme 1. Retrosynthetic analysis of buflavine 1.

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Scheme 2. Synthetic route of buflavine 1.

The eight-membered *N*-heterocyclic ring could be simply formed by a cascade of Pictet–Spengler type reaction then *N*-methylation under the reaction conditions employed. The biarylethylamine precursor **5** is derived from biarylacetonitrile **4** which is easily prepared via Suzuki–Miyaura cross coupling of the commercially available 3,4-dimethoxyphenylboronic acid **2** and *o*-bromophenyl acetonitrile **3** (Scheme 2).

Suzuki-Miyaura cross coupling⁶ of 2 and 3 using 3 mol% Pd(PPh₃)₄ and K₂CO₃ in a mixture of toluene, ethanol and H₂O (3:3:2) gave the expected biarylacetonitrile 4 in 96% yield. Conversion of biarylacetonitrile 4 to the key intermediate, biarylethylamine 5, was accomplished by reduction with NaBH₄ in the presence of cobalt chloride hexahydrate⁷ in 80% yield. We have previously exploited the tandem reaction of the Pictet-Spengler reaction^{8a-d} and Eschweiler-Clarke reaction⁹ in the synthesis of various alkaloids by performing the reaction in formic acid. 10a-c It was gratifying to find that the formation of the eight-membered N-heterocyclic ring and N-methylation could indeed be carried reacting the biarylethylamine paraformaldehyde in formic acid. To complete the ring formation and suppress the non-cyclized N,N-dimethylbiarylethylamine product, the mixture of biarylethylamine 5 and 5 equiv. of paraformaldehyde in formic acid was first stirred at room temperature for 24 h and subsequently an additional 5 equiv. of paraformaldehyde was added and the mixture was then heated at reflux for another 24 h to give buflavine 1 as a viscous oil in 54% yield accompanied by biarylalcohol-amine 6¹¹ in 5% yield after preparative thin layer chromatography (10% MeOH–CH₂Cl₂). The biarylalcohol-amine 6 was presumably derived from electrophilic substitution of the aryl ring with paraformaldehyde in parallel with N,N-dimethylation. The spectroscopic data of our synthetic compound corresponded well with those reported for the natural product. As far as we are aware, this is the first report on the successful application of the Pictet-Spengler type reaction in the formation of an eight-membered heterocyclic ring.

In conclusion, we have developed a concise and highly efficient route for the synthesis of buflavine 1 in three steps in 44% overall yield from commercially available starting materials using a Suzuki–Miyaura cross coupling to form a biaryl ring in conjunction with a cascade of Pictet–Spengler then *N*-methylation of the biarylethylamine in a one-pot reaction to construct the eight-membered heterocyclic ring.

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- 11. All compounds were fully characterized. Biarylacetonitrile (4) pale yellow crystals, mp (EtOAc) 87.0-87.5°C. IR (KBr) 2250 cm⁻¹. 1 H NMR (200 MHz, CDCl₃) δ 3.64 (s, 2H), 3.90, 3.94 (2s, 2×3H), 6.85 (broad s, 2H), 6.95 (d, 1H, J = 8.8 Hz), 7.26 - 7.44 (m, 3H), 7.46 - 7.56 (m, 1H). ¹³C NMR (50 MHz, CDCl₃) δ 22.02, 55.92, 111.31, 112.30, 118.34, 121.11, 127.94, 128.11, 128.98, 130.45, 132.51, 141.75, 148.66, 148.93. EIMS (m/z) % relative intensity) 254 (M++1, 25), 253 (M+, 100), 210 (54), 192 (16), 182 (15), 180 (11), 166 (11), 165 (17). HRMS (FAB+) calcd for C₁₆H₁₆NO₂, 254.1181. Found 254.1179. Anal. calcd for C₁₆H₁₅NO₂: C, 75.87; H, 5.97; N, 5.53. Found: C, 75.51; H, 5.74; N, 5.26%. Biarylethylamine (5) viscous oil, mp (HCl salt) 209–210°C, mp (oxalate salt) 148–149°C. IR (neat) 3345, 2934, 1605, 1585, 1520, 1485, 1246, 1027 cm⁻¹. ¹H NMR (200 MHz, CDCl₃) δ 1.43 (s, 2H), 2.78 (s, 4H), 3.88, 3.93 (2s, 2×3H), 6.80-6.95 (m,

3H), 7.2–7.4 (m, 4H). 13 C NMR (50 MHz, CDCl₃) δ 55.43, 110.62, 112.33, 120.88, 125.59, 126.79, 129.09, 129.81, 134.0, 136.59, 141.60, 147.62, 148.09. EIMS (*m/z*, % relative intensity) 258 (M++1, 6), 257 (M+, 24), 240 (45), 228 (100), 213 (17), 209 (12). HRMS (FAB+) calcd for C₁₆H₂₀NO₂, 258.1494. Found 258.1496. Buflavine 1 viscous oil, mp (oxalate salt) 194.0-194.5°C. IR (neat) 2931, 1606, 1520, 1452, 1357 cm⁻¹. ¹H NMR (200 MHz, CDCl₃) δ 2.49 (s, 3H), 2.49–2.80 (m, 3H), 3.07 (d, 1H, J = 13.5 Hz), 3.26 (t, 1H, J = 9.5 Hz), 3.54 (d, 1H, J = 13.5 Hz) Hz), 3.90, 3.96 (2s, 2×3H), 6.80 (s, 1H), 6.91 (s, 1H), 7.31 (m, 4H). 13 C NMR (50 MHz, CDCl₃) δ 32.45, 45.82, 55.89, 58.24, 58.62, 112.06, 113.51, 126.08, 127.88, 129.01, 129.43, 129.58, 132.92, 139.96, 141.19, 147.84, 148.34. EIMS $(m/z, \% \text{ relative intensity}) 284 (M^++1, 37), 283$ (M⁺, 99.8), 268 (100), 240 (67), 225 (71), 197 (76), 179 (47), 178 (28), 165 (38), 153 (22), 152 (27). HRMS (FAB+) calcd for C₁₈H₂₂NO₂, 284.1651. Found 284.1649. Biarylalcohol-amine 6 viscous oil, IR (neat) 3364, 2936, 1607, 1515, 1465 cm⁻¹. 1 H NMR (200 MHz, CDCl₃) δ 2.00 (s, 6H), 2.40-2.9 (m, 4H), 3.20 (broad s, 1H), 3.85, 3.95 (2s, $2\times3H$), 4.24 (d, 1H, J=12 Hz),), 4.48 (d, 1H, J = 12 Hz), 6.67 (s, 1H), 7.08 (s, 1H), 7.20–7.38 (m, 4H). ¹³C NMR (50 MHz, CDCl₃) δ 30.15, 44.74, 55.95, 56.01, 60.81, 62.49, 112.37, 112.94, 125.94, 127.73, 129.57, 130.29, 131.92, 132.54, 138.52, 140.01, 147.76, 148.35. EIMS (m/z, %) relative intensity) 316 $(M^++1, 26)$, 315 $(M^+, 14), 270 (56), 239 (100), 240 (28), 238 (23), 211 (23),$ 196 (27), 195 (24), 181 (33), 165 (48), 153 (33), 152 (44), 58 (88). HRMS (FAB+) calcd for C₁₉H₂₆NO₃, 316.1913. Found 316.1909.